=> fil reg FILE 'REGISTRY' ENTERED AT 07:12:30 ON 27 FEB 2007 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2007 American Chemical Society (ACS)

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STRUCTURE FILE UPDATES: 26 FEB 2007 HIGHEST RN 923293-79-2 DICTIONARY FILE UPDATES: 26 FEB 2007 HIGHEST RN 923293-79-2

New CAS Information Use Policies, enter HELP USAGETERMS for details.

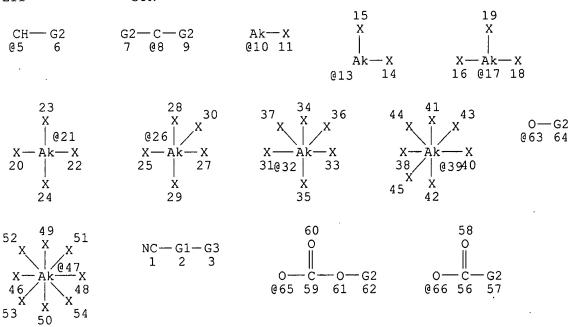
TSCA INFORMATION NOW CURRENT THROUGH June 30, 2006

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

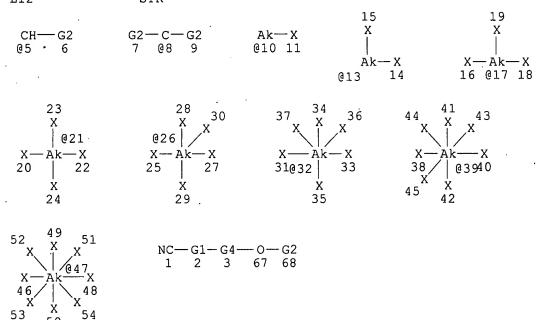
=> d sta que 116 L11 S'



VAR G1=CH2/5/8
VAR G2=AK/10/13/17/21/26/32/39/47
VAR G3=63/66/65
NODE ATTRIBUTES:
CONNECT IS M1 RC AT 47
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 63

STEREO ATTRIBUTES: NONE L12 STR



VAR G1=CH2/5/8
VAR G2=AK/10/13/17/21/26/32/39/47
REP G4=(1-2) CH2
NODE ATTRIBUTES:
CONNECT IS M1 RC AT 47
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 54

STEREO ATTRIBUTES: NONE

L14 SCR 2043 OR 1838 OR 1993 OR 2021 OR 2016 OR 2026 OR 1918 O R 2039 OR 2050 OR 2049 OR 2053 OR 2052 OR 2051 OR 2054 OR 2127 L16 724 SEA FILE=REGISTRY CSS FUL (L11 OR L12) NOT L14

100.0% PROCESSED 10659 ITERATIONS ( 26 INCOMPLETE) 724 ANSWERS SEARCH TIME: 00.06.41

=> fil hcaplus FILE 'HCAPLUS' ENTERED AT 07:12:46 ON 27 FEB 2007 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS)

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jan delaval - 27 february 2007

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FILE COVERS 1907 - 27 Feb 2007 VOL 146 ISS 10 FILE LAST UPDATED: 26 Feb 2007 (20070226/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d 164 bib abs hitind hitstr retable tot

```
L64 ANSWER 1 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN
```

AN 2005:497321 HCAPLUS

DN 143:29529

TI Nonaqueous electrolytes having an extended temperature range for battery applications

IN Sun, Luying

PA USA

SO U.S. Pat. Appl. Publ., 17 pp. CODEN: USXXCO

DT Patent

LA · English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI PRAI	US 2005123835 US 2003-731268	A1	20050609 20031209	US 2003-731268	20031209 <

OS MARPAT 143:29529

AB The present invention discloses non-aqueous electrolytes having an extended temperature range for **battery** applications. The electrolyte comprises an electrolyte salt, e.g., LiPF6, a first non-aqueous solvent, and a second non-aqueous solvent. The electrolyte of the present invention has higher ionic conductivity, lower f.p., and lower vapor pressure at high temperature than com.

electrolytes. These non-aqueous electrolytes can be used, for example, in lithium-ion batteries. Methods of making lithium-ion batteries are also described.

IC ICM H01M0010-40

ICS H01M0004-52; H01M0004-50; H01M0004-58

INCL 429326000; 429330000; 429339000; 429231300; 429231100; 429223000; 429221000; 429224000; 429231800

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 72, 76

ST battery nonaq electrolyte extended temp range

IT Electrochromic devices

Sensors

(electrolyte; nonaq. electrolytes having extended temperature range for battery applications)

IT Secondary batteries

(lithium; nonaq. electrolytes having extended temperature range for battery applications)

IT Battery electrolytes

```
Electrolytic capacitors
     Fuel cell electrolytes
     Ionic conductivity
        (nonaq. electrolytes having extended temperature range for battery
        applications)
IT
     Carbonaceous materials (technological products)
     Coke
     Esters, uses
     Ethers, uses
     RL: DEV (Device component use); USES (Uses)
        (nonaq. electrolytes having extended temperature range for battery
        applications)
ΙT
     Sulfonic acids, uses
     RL: DEV (Device component use); USES (Uses)
        (perfluoro, lithium salt; nonaq. electrolytes having extended temperature
        range for battery applications)
IT
     Perfluoro compounds
     RL: DEV (Device component use); USES (Uses)
        (sulfonic acids, lithium salt; nonaq. electrolytes having extended
        temperature range for battery applications)
ΙT
     96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate
     108-32-7, Propylene carbonate 110-67-8,
     3-Methoxypropionitrile 463-79-6D, Carbonic acid, ester, cyclic
     463-79-6D, Carbonic acid, ester, linear 616-38-6,
     Dimethyl carbonate 623-53-0, Ethyl methyl carbonate
                                       1656-48-0 1738-36-9,
     1001-55-4, 2-Acetoxyacetonitrile
     Methoxyacetonitrile 2141-62-0, 3-Ethoxypropionitrile
     7782-42-5, Graphite, uses 7791-03-9, Lithium perchlorate
     12031-65-1, Lithium nickel oxide (LiNiO2) 12057-17-9,
     Lithium manganese oxide (LiMn2O4) 12190-79-3, Cobalt lithium
     oxide (CoLiO2) 14283-07-9, Lithium tetrafluoroborate
     15365-14-7, Iron lithium phosphate felipo4 18804-04-1,
     uses 21324-40-3, Lithium hexafluorophosphate 29935-35-1
      Lithium hexafluoroarsenate 56756-91-3 62957-60-2,
     Ethoxyacetonitrile 90076-65-6 260362-83-2
     311346-25-5, Cobalt lithium nickel oxide (Co0.1-0.9LiNi0.1-0.902)
     852995-04-1
     RL: DEV (Device component use); USES (Uses)
        (nonaq. electrolytes having extended temperature range for battery
        applications)
ΙT
     96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate
    108-32-7, Propylene carbonate 110-67-8,
     3-Methoxypropionitrile 463-79-6D, Carbonic acid, ester, cyclic
     616-38-6, Dimethyl carbonate 623-53-0, Ethyl methyl
     carbonate 1001-55-4, 2-Acetoxyacetonitrile 1738-36-9,
     Methoxyacetonitrile 2141-62-0, 3-Ethoxypropionitrile
     7782-42-5, Graphite, uses 7791-03-9, Lithium perchlorate
     12031-65-1, Lithium nickel oxide (LiNiO2) 12057-17-9,
     Lithium manganese oxide (LiMn2O4) 12190-79-3, Cobalt lithium
     oxide (CoLiO2) 14283-07-9, Lithium tetrafluoroborate
     15365-14-7, Iron lithium phosphate felipo4 18804-04-1,
     uses 21324-40-3, Lithium hexafluorophosphate 29935-35-1
     , Lithium hexafluoroarsenate 56756-91-3 62957-60-2,
     Ethoxyacetonitrile 90076-65-6 260362-83-2
     311346-25-5, Cobalt lithium nickel oxide (Co0.1-0.9LiNi0.1-0.902)
     852995-04-1
     RL: DEV (Device component use); USES (Uses)
        (nonaq. electrolytes having extended temperature range for battery
        applications)
     96-49-1 HCAPLUS
RN
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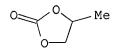
CN 1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)

RN 105-58-8 HCAPLUS

CN Carbonic acid, diethyl ester (CA INDEX NAME)

RN 108-32-7 HCAPLUS

CN 1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)



RN 110-67-8 HCAPLUS

CN Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)

 $MeO-CH_2-CH_2-CN$ 

Group D

RN 463-79-6 HCAPLUS

CN Carbonic acid (7CI, 8CI, 9CI) (CA INDEX NAME)

RN 616-38-6 HCAPLUS

CN Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 623-53-0 HCAPLUS

CN Carbonic acid, ethyl methyl ester (7CI, 8CI, 9CI) (CA INDEX NAME)

RN 1001-55-4 HCAPLUS

CN Acetonitrile, (acetyloxy) - (9CI) (CA INDEX NAME)

Aco-CH2-CN

group ?

RN 1738-36-9 HCAPLUS

CN Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

 $H_3C-O-CH_2-C=N$ 

group A

RN 2141-62-0 HCAPLUS

CN Propanenitrile, 3-ethoxy- (9CI) (CA INDEX NAME)

 ${\tt EtO-CH_2-CH_2-CN}$ 

Group D

EOJ

RN 7782-42-5 HCAPLUS

CN Graphite (CA INDEX NAME)

C

RN 7791-03-9 HCAPLUS

CN Perchloric acid, lithium salt (8CI, 9CI) (CA INDEX NAME)

● T.i

RN 12031-65-1 HCAPLUS

CN Lithium nickel oxide (LiNiO2) (6CI, 8CI, 9CI) (CA INDEX NAME)

Component	    !	Ratio	 	Component Registry Number
	+			
0		2	1	17778-80-2
Ni	1	1	1	7440-02-0
Li	İ	1	i	7439-93-2

RN 12057-17-9 HCAPLUS

CN Lithium manganese oxide (LiMn2O4) (6CI, 7CI, 9CI) (CA INDEX NAME)

Component	Ratio	1	Component
_	1	F	Registry Number
	+=================	+===	
0	4	1	17778-80-2

Mn | 2 | 7439-96-5 Li | 1 | 7439-93-2

RN 12190-79-3 HCAPLUS

CN Cobalt lithium oxide (CoLiO2) (9CI) (CA INDEX NAME)

Component		Ratio	Component Registry Number
==========	==+=	=======================================	
0	- 1	2	17778-80-2
Co	- 1	1	7440-48-4
Li	- 1	1	7439-93-2

RN 14283-07-9 HCAPLUS

CN Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

● Li+

RN 15365-14-7 HCAPLUS CN Phosphoric acid, iron(2+) lithium salt (1:1:1) (9CI) (CA INDEX NAME)

● Fe(II)

• Li

RN 18804-04-1 HCAPLUS .
CN Carbonic acid, 1-cyano-1-methylethyl methyl ester (9CI) (CA INDEX NAME)

group C 38.

RN 21324-40-3 HCAPLUS CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

gon dains 1, 2, 19

● Li+

RN 29935-35-1 HCAPLUS CN Arsenate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

● T.i +

RN 56756-91-3 HCAPLUS
CN Propanenitrile, 2-(acetyloxy)-2-methyl- (9CI) (CA INDEX NAME)

Grang B

RN 62957-60-2 HCAPLUS

CN Acetonitrile, ethoxy- (6CI, 9CI) (CA INDEX NAME)

group A

Eto-CH2-CN

RN 90076-65-6 HCAPLUS

● Li

RN 260362-83-2 HCAPLUS

CN Carbonic acid, 2-cyanoethyl methyl ester (9CI) (CA INDEX NAME)

MeO-C-O-CH<sub>2</sub>-CH<sub>2</sub>-CN

MeO-C-O-CH<sub>2</sub>-CH<sub>2</sub>-CN

Thom dain 4

RN 311346-25-5 HCAPLUS

CN Cobalt lithium nickel oxide (Co0.1-0.9LiNi0.1-0.902) (9CI) (CA INDEX NAME)

Component		Ratio	Component   Registry Number
	+=		+
0	- 1	. 2	17778-80-2
Co	1	0.1 - 0.9	7440-48-4
Ni	1	0.1 - 0.9	7440-02-0
Li	- 1	1	7439-93-2

RN 852995-04-1 HCAPLUS

CN Carbonic acid, cyanomethyl methyl ester (9CI) (CA INDEX NAME)

L64 ANSWER 2 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2004:286861 HCAPLUS

DN .140:290041

TI Electrolyte composition having improved aluminum anticorrosive properties

IN Exnar, Ivan; Di Censo, Davide

```
Xoliox S. A., Switz.
PA
SO
     Eur. Pat. Appl., 22 pp.
     CODEN: EPXXDW
DT
     Patent
     English
LA
FAN: CNT 1
                         KIND
                                            APPLICATION NO.
                                                                    DATE
     PATENT NO.
     -----
                                            -----
                                20040407 EP 2002-405848
     EP 1406336
                          Α1
                                                                    20021001 <--
PI
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK
PRAI EP 2002-405848
                                20021001 <--
     The invention relates to an electrolyte composition for use in an electrochem.
     battery having an aluminum current collector, the composition
     comprising an imide salt and a nitrile-based solvent.
IC
     ICM H01M0010-40
     ICS H01M0006-18; H01M0004-66
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
     Section cross-reference(s): 72
     battery electrolyte compn improved aluminum anticorrosive
     property
IT
     Amperometry
       Battery electrolytes
     Corrosion prevention
     Plasticizers
        (electrolyte composition having improved aluminum anticorrosive properties)
     96-48-0, \gamma-Butyrolactone 96-49-1, Ethylene carbonate
IT
     105-58-8, Diethyl carbonate 108-32-7, Propylene
     carbonate 109-99-9, Thf, uses 110-71-4 616-38-6, Dimethyl
     carbonate 623-53-0, Ethyl methyl carbonate 646-06-0,
     1,3-Dioxolane 1738-36-9, Methoxyacetonitrile
                                                    9003-05-8,
                                                   9003-21-8,
     Polyacrylamide 9003-20-7, Polyvinylacetate
     Polymethylacrylate 9003-39-8, Polyvinylpyrrolidone 12031-95-7,
     Lithium titanium oxide li4ti5012 12190-79-3, Cobalt lithium
     oxide colio2 24937-79-9, Polyvinylidene fluoride 25322-68-3, Peo
     26809-02-9, Polyacetonitrile 57619-91-7, Polytetraethylene glycol
     diacrylate 73506-93-1, Diethoxyethane 90076-65-6, Lithium
     bis(trifluoromethylsulfonyl)imide 132843-44-8, Lithium
     bis (perfluoroethylsulfonyl) imide
     RL: DEV (Device component use); USES (Uses)
        (electrolyte composition having improved aluminum anticorrosive properties)
ΙT
     96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate
     108-32-7, Propylene carbonate 616-38-6, Dimethyl
     carbonate 623-53-0, Ethyl methyl carbonate 646-06-0,
     1,3-Dioxolane 1738-36-9, Methoxyacetonitrile 12031-95-7
     , Lithium titanium oxide li4ti5012 12190-79-3, Cobalt lithium
     oxide colio2 90076-65-6, Lithium bis(trifluoromethylsulfonyl)imi
     de 132843-44-8, Lithium bis(perfluoroethylsulfonyl)imide
     RL: DEV (Device component use); USES (Uses)
        (electrolyte composition having improved aluminum anticorrosive properties)
RN
     96-49-1 HCAPLUS
     1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)
CN
```



RN 105-58-8 HCAPLUS

CN Carbonic acid, diethyl ester (CA INDEX NAME)

RN 108-32-7 HCAPLUS

CN 1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)

RN 616-38-6 HCAPLUS

CN Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 623-53-0 HCAPLUS

CN Carbonic acid, ethyl methyl ester (7CI, 8CI, 9CI) (CA INDEX NAME)

RN 646-06-0 HCAPLUS

CN 1,3-Dioxolane (6CI, 8CI, 9CI) (CA INDEX NAME)



RN 1738-36-9 HCAPLUS

CN Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

 $H_3C-O-CH_2-C \equiv N$ 

group A

RN 12031-95-7 HCAPLUS

CN Lithium titanium oxide (Li4Ti5O12) (CA INDEX NAME)

Component	 	Ratio .		Component Registry Number
	==+==		+=	===============
0	1	12	1	17778-80-2
Ti ·	- 1	5	1	7440-32-6
Li	ĺ	4	ĺ	7439-93-2

jan delaval - 27 february 2007

RN 12190-79-3 HCAPLUS

CN Cobalt lithium oxide (CoLiO2) (9CI) (CA INDEX NAME)

Component	   	Ratio	Component Registry Number
	+-	·~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
0	- 1	2	17778-80-2
Co	-1	1 !	7440-48-4
Li	Ì	. 1	7439-93-2

RN 90076-65-6 HCAPLUS

CN Methanesulfonamide, 1,1,1-trifluoro-N-[(trifluoromethyl)sulfonyl]-, lithium salt (9CI) (CA INDEX NAME)

# • Li

RN 132843-44-8 HCAPLUS

CN Ethanesulfonamide, 1,1,2,2,2-pentafluoro-N-[(pentafluoroethyl)sulfonyl]-, lithium salt (9CI) (CA INDEX NAME)

$$F_3C-CF_2-S-NH-S-CF_2-CF_3$$

# • Li

L64 ANSWER 3 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2000:624975 HCAPLUS

DN 133:225554

TI Nonaqueous electrolyte solutions containing cyanoethyl compounds and nonaqueous (lithium) secondary batteries

IN Toriida, Masahiro; Omi, Katsuhiko; Tan, Hiroaki

PA Mitsui Chemical Industry Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	* - ·				
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	JP 2000243444	Α	20000908	JP 1999-41104	19990219 <
DDAT	TD 1999-41104		10000210	/ <b>_</b> _	

```
OS
     MARPAT 133:225554
     The solns. are nonaq. solvents containing RO(R10) nCH2CH2CN (R = H, C1-10
AB
     hydrocarbon, cyanoethyl; R1 = C1-4 alkylene; n = integer or 0-30) and
     electrolytes. The solns. may also contain linear carbonate esters and/or
     cyclic carbonate esters given in Markush structures. Secondary
     batteries, especially lithium ion batteries, comprising the
     electrolyte solns. are also claimed. Batteries with excellent
     charge-discharge characteristics and high performance, under loaded
     conditions and low-temperature, are obtained.
     ICM H01M0010-40
IC
     ICS H01M0004-58
CC
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
ST
     lithium secondary battery nonaq electrolyte; electrolyte soln
     cyanoethyl additive secondary battery; cyclic carbonate nonaq
     electrolyte secondary battery; linear carbonate nonaq
     electrolyte secondary battery; carbonate nonaq electrolyte
     secondary battery
ΙT
     Secondary batteries
        (lithium; secondary (lithium) batteries comprising of nonag.
        solvents containing cyanoethyl compds.)
IT
     Battery electrolytes
        (secondary (lithium) batteries comprising of nonaq. solvents
        containing cyanoethyl compds.)
IT
     Lithium alloy, base
     RL: DEV (Device component use); USES (Uses)
        (anode; secondary (lithium) batteries comprising of nonaq.
        solvents containing cyanoethyl compds.)
ΙT
     12190-79-3, HLC 21
     RL: DEV (Device component use); USES (Uses)
        (HLC 21, cathode; secondary (lithium) batteries comprising of
        nonaq. solvents containing cyanoethyl compds.)
     7439-93-2, Lithium, uses 7440-44-0, MCMB 6-28, uses
ΙT
     RL: DEV (Device component use); USES (Uses)
        (anode; secondary (lithium) batteries comprising of nonag.
        solvents containing cyanoethyl compds.)
     105-58-8, Diethyl carbonate 108-32-7, Propylene
IT
     carbonate 110-67-8 616-38-6, Dimethyl carbonate
     623-53-0, Methyl ethyl carbonate 1656-48-0, Bis(2-cyanoethyl)
     ether 2141-62-0
                     3386-87-6 4437-85-8, Butylene
     carbonate
                35633-50-2
     RL: DEV (Device component use); USES (Uses)
        (secondary (lithium) batteries comprising of nonaq. solvents
        containing cyanoethyl compds.)
IT
     12190-79-3, HLC 21
     RL: DEV (Device component use); USES (Uses)
        (HLC 21, cathode; secondary (lithium) batteries comprising of
        nonaq. solvents containing cyanoethyl compds.)
     12190-79-3 HCAPLUS
RN
CN
     Cobalt lithium oxide (CoLiO2) (9CI) (CA INDEX NAME)
```

Component	1	Ratio	appar and a	Component Registry Number
=========	==+==:		====+=	
. 0	- 1	2	. 1	17778-80-2
Co	1	1	1	7440-48-4
Li	1	1	1	7439-93-2

7439-93-2, Lithium, uses 7440-44-0, MCMB 6-28, uses IT RL: DEV (Device component use); USES (Uses) (anode; secondary (lithium) batteries comprising of nonag.

```
solvents containing cyanoethyl compds.)
     7439-93-2 HCAPLUS
RN
CN
     Lithium (CA INDEX NAME)
Li
RN
     7440-44-0 HCAPLUS
CN
     Carbon (CA INDEX NAME)
С
IT
     105-58-8, Diethyl carbonate 108-32-7, Propylene
     carbonate 110-67-8 616-38-6, Dimethyl carbonate
     623-53-0, Methyl ethyl carbonate 2141-62-0
     4437-85-8, Butylene carbonate
     RL: DEV (Device component use); USES (Uses)
        (secondary (lithium) batteries comprising of nonaq. solvents
        containing cyanoethyl compds.)
     105-58-8 HCAPLUS
RN
CN
     Carbonic acid, diethyl ester (CA INDEX NAME)
EtO-C-OEt
RN
     108-32-7 HCAPLUS
     1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)
RN
    110-67-8 HCAPLUS
CN
    Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)
MeO-CH2-CH2-CN
RN
     616-38-6 HCAPLUS
    Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)
CN
MeO-C-OMe
RN
     623-53-0 HCAPLUS
    Carbonic acid, ethyl methyl ester (7CI, 8CI, 9CI) (CA INDEX NAME)
CN
```

```
0
||
MeO-C-OEt
```

RN 2141-62-0 HCAPLUS

CN Propanenitrile, 3-ethoxy- (9CI) (CA INDEX NAME)

EtO-CH2-CH2-CN

RN 4437-85-8 HCAPLUS

CN 1,3-Dioxolan-2-one, 4-ethyl- (9CI) (CA INDEX NAME)

L64 ANSWER 4 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2000:624973 HCAPLUS

DN 133:225552

TI Nonaqueous electrolyte solutions containing cyanoethyl compounds and nońaqueous (lithium) secondary **batteries** 

IN Toriida, Masahiro; Omi, Takehiko; Tan, Hiroaki

PA Mitsui Chemical Industry Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

ran.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	JP 2000243442 JP 1999-41102	Α	20000908 19990219	JP 1999-41102 · <	19990219 <

OS · MARPAT 133:225552

AB The solns. are nonaq. solvents containing ROnCO2CH2CH2CN (R = H, C1-10 hydrocarbon, CH2CH2CN; n = 0, 1) and electrolytes. The solns. may also contain linear carbonate esters and/or cyclic carbonate esters given in Markush structures. Secondary batteries, especially lithium ion batteries, comprising the electrolyte solns. are also claimed.

Batteries with excellent charge-discharge characteristics and high performance, under loaded conditions and low-temperature, are obtained.

IC ICM H01M0010-40

ICS H01M0004-02; H01M0004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST lithium secondary battery nonaq electrolyte; electrolyte soln cyanoethyl additive secondary battery; cyclic carbonate nonaq electrolyte secondary battery; linear carbonate nonaq electrolyte secondary battery; carboxylate nonaq electrolyte secondary battery

IT Secondary batteries

(lithium; secondary (lithium) batteries comprising of nonaq. solvents containing cyanoethyl compds.)

IT Battery electrolytes

(secondary (lithium) batteries comprising of nonaq. solvents containing cyanoethyl compds.)

```
IT
     Lithium alloy, base
     RL: DEV (Device component use); USES (Uses)
        (anode; secondary (lithium) batteries comprising of nonaq.
        solvents containing cyanoethyl compds.)
TT
     12190-79-3, HLC 21
     RL: DEV (Device component use); USES (Uses)
        (HLC 21, cathode; secondary (lithium) batteries comprising of
        nonaq. solvents containing cyanoethyl compds.)
     7439-93-2, Lithium, uses 7440-44-0, MCMB 6-28, uses
IT
     RL: DEV (Device component use); USES (Uses)
        (anode; secondary (lithium) batteries comprising of nonaq.
        solvents containing cyanoethyl compds.)
ΙT
     105-58-8, Diethyl carbonate 108-32-7, Propylene
     carbonate 616-38-6, Dimethyl carbonate 623-53-0,
     Methyl ethyl carbonate 4437-85-8, Butylene carbonate
     20597-73-3, 2-Cyanoethyl propionate 260362-83-2
     RL: DEV (Device component use); USES (Uses)
        (secondary (lithium) batteries comprising of nonaq. solvents
        containing cyanoethyl compds.)
     12190-79-3, HLC 21
IT
     RL: DEV (Device component use); USES (Uses)
        (HLC 21, cathode; secondary (lithium) batteries comprising of
        nonaq. solvents containing cyanoethyl compds.)
     12190-79-3 HCAPLUS
RN
     Cobalt lithium oxide (CoLiO2) (9CI) (CA INDEX NAME)
CN
```

Component		Ratio	1	Component Registry Number
~~~~~~	==+===		====+==	
0	- 1	2		17778-80-2
Co	1	1		7440-48-4
Li	- 1	1	1	7439-93-2

7439-93-2, Lithium, uses 7440-44-0, MCMB 6-28, uses TT RL: DEV (Device component use); USES (Uses) (anode; secondary (lithium) batteries comprising of nonaq. solvents containing cyanoethyl compds.) 7439-93-2 HCAPLUS RN

Lithium (CA INDEX NAME) CN

Li

RN 7440-44-0 HCAPLUS Carbon (CA INDEX NAME) CN

С

ΙT 105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 616-38-6, Dimethyl carbonate 623-53-0, Methyl ethyl carbonate 4437-85-8, Butylene carbonate 260362-83-2 RL: DEV (Device component use); USES (Uses) (secondary (lithium) batteries comprising of nonaq. solvents containing cyanoethyl compds.) RN 105-58-8 HCAPLUS

CN Carbonic acid, diethyl ester (CA INDEX NAME) RN 108-32-7 HCAPLUS

CN 1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)

O Me

RN 616-38-6 HCAPLUS

CN Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

MeO-C-OMe

RN 623-53-0 HCAPLUS

CN Carbonic acid, ethyl methyl ester (7CI, 8CI, 9CI) (CA INDEX NAME)

RN 4437-85-8 HCAPLUS

CN 1,3-Dioxolan-2-one, 4-ethyl- (9CI) (CA INDEX NAME)

O Et

RN 260362-83-2 HCAPLUS

CN Carbonic acid, 2-cyanoethyl methyl ester (9CI) (CA INDEX NAME)

" grows e" - claim 4 (1 extra carbon in chain).

L64 ANSWER 5 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2000:399134 HCAPLUS

DN 133:20143

TI Nonaqueous electrolyte solutions and secondary nonaqueous electrolyte

IN Hayashi, Takeshi; Tan, Hiroaki

PA Mitsui Petrochemical Industries, Co., Ltd., Japan

```
· $0
     Jpn. Kokai Tokkyo Koho, 7 pp.
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
FAN.CNT 1
                         KIND
                                 DATE
                                             APPLICATION NO.
                                                                    DATE
     PATENT NO.
                          ____
                                 -----
PΙ
     JP 2000164249
                          Α
                                 20000616
                                           JP 1998-336632
                                                                    19981127 <--
PRAI JP 1998-336632
                                 19981127 <--
OS
     MARPAT 133:20143
AB
     The electrolyte solns. contain an electrolyte and a nonaq. solvent containing
     a F-containing cyanoethyl ether X(OR)n)CH2CH2CN, where X = F substituted C1-10
     hydrocarbon group, R =C2-4 alkylene group, and n = 0-30. The salt is
     selected from LiPF6, LiBF4, and Li salts of S containing organic acids. The
     batteries are secondary Li batteries.
IC
     ICM H01M0010-40
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
CC
     secondary lithium battery electrolyte solvent; lithium
     battery electrolyte solvent fluorinated cyanoethyl ether
IT
     Battery electrolytes
         (nonaq. solvent mixts. containing fluorinated cyanoethyl ethers for
        electrolytes in secondary lithium batteries)
ΙT
     96-49-1, Ethylene carbonate 353-18-4 616-38-6,
     Dimethyl carbonate 21324-40-3, Lithium hexafluorophosphate
     85169-02-4 272128-06-0
                             272128-07-1
                                           272128-08-2 272128-09-3
     272128-10-6
     RL: DEV (Device component use); USES (Uses)
         (nonag. solvent mixts. containing fluorinated cyanoethyl ethers for
        electrolytes in secondary lithium batteries)
IT
     96-49-1, Ethylene carbonate 353-18-4 616-38-6,
     Dimethyl carbonate 21324-40-3, Lithium hexafluorophosphate
     272128-06-0
     RL: DEV (Device component use); USES (Uses)
         (nonaq. solvent mixts. containing fluorinated cyanoethyl ethers for
        electrolytes in secondary lithium batteries)
     96-49-1 HCAPLUS
RN
     1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)
CN
RN
     353-18-4 HCAPLUS
CN
     Propanenitrile, 3-(2-fluoroethoxy)- (9CI) (CA_INDEX_NAME)
FCH2-CH2-O-CH2-CH2-CN
```

0 ||. MeO- C- OMe

RN

CN

616-38-6 HCAPLUS

Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

```
21324-40-3 HCAPLUS
RN
```

CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

#### Li+

272128-06-0 HCAPLUS RN

CN Propanenitrile, 3-(2,2,2-trifluoroethoxy)- (9CI) (CA INDEX NAME)

F3C-CH2-O-CH2-CH2-CN

ANSWER 6 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

2000:254779 HCAPLUS

132:267606

Organic electrolyte solutions for batteries and capacitors TI

Nishikawa, Satoshi; Bessho, Shinji

PA Sunstar Engineering, Inc., Japan; Uni Sunstar Bv

SO Jpn. Kokai Tokkyo Koho, 5 pp. CODEN: JKXXAF

DT Patent

Japanese LA

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
ΡI	JP 2000113906	Α	20000421	JP 1998-282341	19981005 <	
PRAI	JP 1998-282341		19981005	<		
OS	MARPAT 132:267606					
W D.	mb 1 + 1 - + 1		1		0 \ = 0N	

The electrolyte solns.use a solvent containing ≥1 of RCOO(CH2)aCN, AB where R = H or C1-3 alkyl or alkoxy group, a = integer 1-3. The electrolyte is a Li salt or a tetraalkyl quaternary ammonium or phosphonium salt.

IC ICM H01M0010-40

ICS C07C0255-14; H01G0009-038; H01G0009-035; H01M0006-16

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) .

battery electrolyte solvent nitrile ester; capacitor electrolyte STsolvent nitrile ester

ΙT Capacitors

> (double layer; electrolyte solns. containing carboxylic acid nitrile esters for batteries and elec. capacitors)

ΙT Battery electrolytes

(electrolyte solns. containing carboxylic acid nitrile esters for batteries and elec. capacitors)

5325-93-9, 2-Cyanoethyl acetate 20597-73-3, 2-Cyanoethyl propionate ΙT

**21324-40-3**, Lithium hexafluorophosphate 154119-71-8 **260362-83-2** 

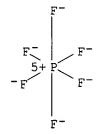
RL: TEM (Technical or engineered material use); USES (Uses) (electrolyte solns. containing carboxylic acid nitrile esters for batteries and elec. capacitors)

IT 21324-40-3, Lithium hexafluorophosphate 260362-83-2

RL: TEM (Technical or engineered material use); USES (Uses) (electrolyte solns. containing carboxylic acid nitrile esters for batteries and elec. capacitors)

RN 21324-40-3 HCAPLUS

CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)



● Li+

RN 260362-83-2 HCAPLUS

CN Carbonic acid, 2-cyanoethyl methyl ester (9CI) (CA INDEX NAME)

"Group C"
(extra chrbon)

L64 ANSWER 7 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2000:166259 HCAPLUS

DN 132:210209

TI Secondary nonaqueous-electrolyte **batteries** with electrolytes containing cyanoethoxy compounds

IN Kobayashi, Aya; Izuchi, Shuichi

PA Yuasa Battery Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	0.11					
	PATENT NO.	KIND DATE		APPLICATION NO.	DATE	
PI	JP 2000\077096	A	20000314	JP 1998-244674	19980831 <	
PRAI	JP 1998'∸244674		19980831	<		
os	MARPAT 132:210209					

AB Claimed batteries are equipped with electrolytes containing cyanoethoxy compds. R(OC2H4CN)n (n = 1-4; R = CmH2m+2-n, CmH2m+2-n(OC2H4)p, CmH2m+2-nCO, or CmH2m+2-nOCO; m = 1-3; p = 1-4) as nonag. solvents for Li salts. Optionally, the batteries are

```
equipped with gelled polymer electrolytes. The batteries have
     long cycle life at low temperature
IC
     ICM H01M0010-40
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
CC
     cyanoethoxy compd nonaq electrolyte solvent battery; lithium
     battery electrolyte solvent cyanoethoxy compd
IT
     Secondary batteries
        (lithium; nonaq. batteries with electrolytes containing
        cyanoethoxy compds. for long cycle life at low temperature)
ΙT
     Battery electrolytes
        (nonaq. batteries with electrolytes containing cyanoethoxy
        compds. for long cycle life at low temperature)
ΙT
     Polyoxyalkylenes, uses
     RL: DEV (Device component use); USES (Uses)
        (trifunctional acrylates, lithium complexes, gelled electrolytes;
        nonaq. batteries with electrolytes containing cyanoethoxy compds.
        for long cycle life at low temperature)
     14283-07-9, Lithium tetrafluoroborate
TT
     RL: DEV (Device component use); USES (Uses)
        (electrolytes; nonaq. batteries with electrolytes containing
        cyanoethoxy compds. for long cycle life at low temperature)
IT
     25322-68-3D, Polyethylene glycol, trifunctional acrylates, lithium
     complexes
     RL: DEV (Device component use); USES (Uses)
        (gelled electrolytes; nonaq. batteries with electrolytes
        containing cyanoethoxy compds. for long cycle life at low temperature)
IT
     96-48-0, \gamma-Butyrolactone 96-49-1, Ethylene carbonate
     108-32-7, Propylene carbonate 110-67-8 1656-48-0,
     Bis-2-cyanoethyl ether 2141-62-0 3386-87-6
                                                    5325-93-9
     20597-73-3
                  32846-35-8, Bis 2-cyanoethyl carbonate · 35633-51-3
     260362-83-2
     RL: DEV (Device component use); USES (Uses)
        (solvents; nonaq. batteries with electrolytes containing
        cyanoethoxy compds. for long cycle life at low temperature)
ΙT
     14283-07-9, Lithium tetrafluoroborate
     RL: DEV (Device component use); USES (Uses)
        (electrolytes; nonaq. batteries with electrolytes containing
        cyanoethoxy compds. for long cycle life at low temperature)
RN
     14283-07-9 HCAPLUS
CN
     Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)
```

● Li+

IT 96-49-1, Ethylene carbonate 108-32-7, Propylene
 carbonate 110-67-8 2141-62-0 260362-83-2
 RL: DEV (Device component use); USES (Uses)
 (solvents; nonaq. batteries with electrolytes containing
 cyanoethoxy compds. for long cycle life at low temperature)
RN 96-49-1 HCAPLUS

1,3-Dioxolan-2-one (9CI) CN (CA INDEX NAME)

RN 108-32-7 HCAPLUS

1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)

110-67-8 HCAPLUS RN

Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME) CN

 $MeO-CH_2-CH_2-CN$ 

2141-62-0 HCAPLUS RN

CN Propanenitrile, 3-ethoxy- (9CI) www.bna

EtO-CH2-CH2-CN

RN 260362-83-2 HCAPLUS

Carbonic acid, 2-cyanoethyl methyl ester (9CI) (CA INDEX NAME)

$$0 \\ || \\ MeO-C-O-CH_2-CH_2-CN$$

L64 ANSWER 8 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

ΑN 2000:144319 HCAPLUS

DN 132:183113

Secondary nonaqueous electrolyte batteries TI

Tabuchi, Toru; Aoki, Takashi; Nakamitsu, Kazuhiro; Mizutani, Minoru ΙN

Japan Storage Battery Co., Ltd., Japan; GS Melcotec K. K. PΑ

Jpn. Kokai Tokkyo Koho, 7 pp. SO

CODEN: JKXXAF

DTPatent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2000 067913	A	20000303	JP 1998-305833	19981027 <
PRAI	JP 1998-159629	A	19980608	<	
70 10	The betteries was	2 2222	Ti1+ -	lestualista calistica con	taining a

AΒ The batteries use a nonaq. Li salt electrolyte solution containing a cyano group containing ether or glycol and a carbonate ester.

IC ICM H01M0010-40

52-2 (Electrochemical, Radiational, and Thermal Energy Technology) CC

ST secondary lithium battery electrolyte soln compn; lithium battery electrolyte cyano ether carbonate ester; cyano glycol carbonate ester lithium battery electrolyte

IT Battery electrolytes

(electrolyte solvent mixts. containing cyano ethers or cyano glycols and carbonate esters for secondary lithium **batteries**)

IT 96-49-1, Ethylene carbonate 110-67-8 623-53-0,

Ethyl methyl carbonate 3386-87-6 21324-40-3, Lithium

hexafluorophosphate

RL: DEV (Device component use); USES (Uses)

(electrolyte solvent mixts. containing cyano ethers or cyano glycols and carbonate esters for secondary lithium **batteries**)

IT 96-49-1, Ethylene carbonate 110-67-8 623-53-0,

Ethyl methyl carbonate 21324-40-3, Lithium hexafluorophosphate

RL: DEV (Device component use); USES (Uses) .

(electrolyte solvent mixts. containing cyano ethers or cyano glycols and carbonate esters for secondary lithium **batteries**)

RN 96-49-1 HCAPLUS

CN 1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)

000

RN 110-67-8 HCAPLUS

CN Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)

MeO-CH2-CH2-CN

Group D

RN 623-53-0 HCAPLUS

CN Carbonic acid, ethyl methyl ester (7CI, 8CI, 9CI) (CA INDEX NAME)

| || MeO-- C-- OEt

RN 21324-40-3 HCAPLUS

CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

### ● T.1 +

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L64 ANSWER 9 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN
ΑN
     1999:737130 HCAPLUS
DN
     131:325078
ΤI
     Primary or secondary electrochemical generator
ΙN
     Gratzel, Michael; Sugnaux, Francois R.; Pappas, Nicholas
PA
     Ecole Polytechnique Federale De Lausanne (Epfl) Sri, Switz.
     PCT Int. Appl., 29 pp.
SO
     CODEN: PIXXD2
DT
     Patent
LA
     English
FAN.CNT 1
     PATENT NO.
                         KIND
                                DATE
                                            APPLICATION NO.
                                                                    DATE
                         ----
                                19991118
PΙ
     WO 9959218
                         A1
                                            WO 1999-EP3261
                                                                   19990508 <--
         W: CN, JP, US
         RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE
     EP 1086506
                                            EP 1999-932452
                          Α1
                                20010328
                                                                   19990508 <--
         R: CH, DE, FR, GB, LI, NL, IE
PRAI EP 1998-810431
                         Α
                                19980512
                                          <--
     WO 1999-EP3261
                          W
                                19990508 <--
AB
     A high power d. and high capacity primary or secondary electrochem.
     generator has at least one electrode composed of an elec. active solid
     material, the electrode having a mesoporous texture forming a
     bi-continuous junction of large sp. surface area with the electrolyte.
     The specific morphol. of the electroactive material permits high rates of
     ion insertion in the solid while allowing for rapid ion transport in
     electrolyte present in the porous space of the electrode. Specific
     methods for preparation of the electrode are disclosed, in particular the
     control of the electrode morphol. by use of surfactant assemblies such as
     surfactant micelles exerting a templating effect during the chemical
     synthesis of the electroactive material.
IC
     ICM H01M0010-40
     ICS H01M0004-48; H01M0004-58
     52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
CC
ST
     battery electrode transition metal oxide chalcogenide
ΙT
     Primary batteries
     Secondary batteries
        (lithium; primary or secondary electrochem. generator)
IT
     Battery electrodes
        (primary or secondary electrochem. generator)
ΙT
     7440-44-0, Carbon, uses
                              12597-68-1, Stainless steel, uses
```

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RL: DEV (Device component use); USES (Uses)
         (current collector; primary or secondary electrochem. generator)
ΙT
     7782-42-5, Graphite, uses
     RL: DEV (Device component use); USES (Uses)
         (paper, current collector; primary or secondary electrochem. generator)
IT
     96-48-0 96-49-1, Ethylene carbonate 105-58-8, Diethyl
     carbonate 108-32-7, Propylene carbonate 616-38-6,
     Dimethyl carbonate 646-06-0, Dioxolane 1309-37-1, Iron oxide
                    1312-43-2, Indium oxide
     (Fe2O3), uses
                                               1313-13-9, Manganese dioxide,
     uses
            1313-27-5, Molybdenum trioxide, uses 1313-96-8, Niobium pentoxide
     1314-35-8, Tungsten trioxide, uses 1314-62-1, Vanadium pentoxide, uses 1317-33-5, Molybdenum sulfide mos2, uses 1317-61-9, Iron oxide (Fe304),
     uses 1738-36-9, Methoxyacetonitrile 2923-17-3, Lithium
     trifluoroacetate
                       11113-84-1, Ruthenium oxide 11126-12-8, Iron sulfide
     11129-18-3, Cerium oxide
                                 12039-13-3, Titanium disulfide
                                                                  12055-23-1,
     Hafnium dioxide
                      12067-45-7, Titanium diselenide 12138-09-9, Tungsten
     sulfide ws2
                  12645-46-4, Iridium oxide
                                                13463-67-7, Titania, uses
     14024-11-4, Lithium tetrachloroaluminate 14283-07-9,
     Lithium tetrafluoroborate 18424-17-4, Lithium
     hexafluoroantimonate 21324-40-3, Lithium hexafluorophosphate
     26856-69-9, Methoxypropionitrile
                                        28106-65-2, Tetrafluoropropanol
     29935-35-1, Lithium hexafluoroarsenate 33454-82-9,
     Lithium triflate 37245-92-4, Ruthenium sulfide 39300-70-4,
     Lithium nickel oxide 39457-42-6, Lithium manganese oxide
     52627-24-4, Cobalt lithium oxide 59763-75-6, Tantalum oxide
     66216-18-0 90076-65-6 131344-56-4, Cobalt lithium
     nickel oxide 131651-65-5, 1-Butanesulfonic acid,
     1,1,2,2,3,3,4,4,4-nonafluoro-, lithium salt 132404-42-3
     248588-09-2, Indium lithium manganese sodium oxide
     RL: DEV (Device component use); USES (Uses)
        (primary or secondary electrochem. generator)
IT
     7440-44-0, Carbon, uses
     RL: DEV (Device component use); USES (Uses)
        (current collector; primary or secondary electrochem. generator)
RN
     7440-44-0 HCAPLUS
     Carbon (CA INDEX NAME)
CN
С
IT
     7782-42-5, Graphite, uses
     RL: DEV (Device component use); USES (Uses)
     (paper, current collector; primary or secondary electrochem. generator) 7782-42-5 HCAPLUS
RN
CN
     Graphite (CA INDEX NAME)
С
ΙT
     96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate
     108-32-7, Propylene carbonate 616-38-6, Dimethyl
     carbonate 646-06-0, Dioxolane 1738-36-9,
     Methoxyacetonitrile 2923-17-3, Lithium trifluoroacetate
     14024-11-4, Lithium tetrachloroaluminate 14283-07-9,
     Lithium tetrafluoroborate 18424-17-4, Lithium
     hexafluoroantimonate 21324-40-3, Lithium hexafluorophosphate
     29935-35-1, Lithium hexafluoroarsenate 33454-82-9,
     Lithium triflate 39300-70-4, Lithium nickel oxide
```

39457-42-6, Lithium manganese oxide 52627-24-4, Cobalt lithium oxide 90076-65-6 131344-56-4, Cobalt lithium nickel oxide 131651-65-5, 1-Butanesulfonic acid, 1,1,2,2,3,3,4,4,4-nonafluoro-, lithium salt 132404-42-3 248588-09-2, Indium lithium manganese sodium oxide RL: DEV (Device component use); USES (Uses) (primary or secondary electrochem. generator)

RN 96-49-1 HCAPLUS

CN 1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)

RN 105-58-8 HCAPLUS
CN Carbonic acid, diethyl ester (CA INDEX NAME)

RN 108-32-7 HCAPLUS CN 1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)

RN 616-38-6 HCAPLUS CN Carbonic acid, dimethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 646-06-0 HCAPLUS CN 1,3-Dioxolane (6CI, 8CI, 9CI) (CA INDEX NAME)



RN 1738-36-9 HCAPLUS CN Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

 $_{\mathrm{H_3C-O-CH_2-C}}$ N

RN 2923-17-3 HCAPLUS

group A

CN Acetic acid, trifluoro-, lithium salt (8CI, 9CI) (CA INDEX NAME)

● Li

RN 14024-11-4 HCAPLUS CN Aluminate(1-), tetrachloro-, lithium, (T-4)- (9CI) (CA INDEX NAME)

● Li+

RN 14283-07-9 HCAPLUS CN Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

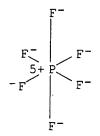
Claim Z

● Li<sup>+</sup>

RN 18424-17-4 HCAPLUS CN Antimonate(1-), hexafluoro-, lithium, (OC-6-11)- (9CI) (CA INDEX NAME)

● Li <sup>4</sup>

RN 21324-40-3 HCAPLUS CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)



Claim 2

1.800.3

■ 1.4 ±

RN 29935-35-1 HCAPLUS CN Arsenate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

● Li<sup>4</sup>

RN 33454-82-9 HCAPLUS CN Methanesulfonic acid, trifluoro-, lithium salt (8CI, 9CI) (CA INDEX NAME)

#### ● T.i

RN 39300-70-4 HCAPLUS

CN Lithium nickel oxide (9CI) (CA INDEX NAME)

Component	1	Ratio		Component Registry Number
==========	==+=:		=+=	
0	1	x	ı	17778-80-2
Ni	1	x	- 1	7440-02-0
Li	1	x		7439-93-2

RN 39457-42-6 HCAPLUS

CN Lithium manganese oxide (9CI) (CA INDEX NAME)

Component		Ratio		Component		
	1		I	Registry Number		
	==+==	<del></del>	==+=			
0		X	1	17778-80-2		
Mn	1	X	- 1	7439-96-5		
Li		X	- 1	7439-93-2		

RN 52627-24-4 HCAPLUS

CN Cobalt lithium oxide (9CI) (CA INDEX NAME)

Component	 	Ratio	1	Component Registry Number
	==+==		===+==	
0	1	x	1	17778-80-2
Co	1	x	1	7440-48-4
Li	1	x	1	7439-93-2

RN 90076-65-6 HCAPLUS

CN Methanesulfonamide, 1,1,1-trifluoro-N-[(trifluoromethyl)sulfonyl]-, lithium salt (9CI) (CA INDEX NAME)

# ● Li

RN 131344-56-4 HCAPLUS

CN Cobalt lithium nickel oxide (CA INDEX NAME)

Component .	1	Ratio		Component Registry Number
=========	==+==	:=========	+=	===========
0	1	x		17778-80-2
Co	1	x		7440-48-4
Ni	- 1	x	1	7440-02-0
Li	- 1	x	-	7439-93-2

RN 131651-65-5 HCAPLUS

CN 1-Butanesulfonic acid, 1,1,2,2,3,3,4,4,4-nonafluoro-, lithium salt (9CI) (CA INDEX NAME)

HO3S- (CF2) 3-CF3

● Li

RN 132404-42-3 HCAPLUS

● Li+

RN 248588-09-2 HCAPLUS

CN Indium lithium manganese sodium oxide (9CI) (CA INDEX NAME)

Component	    !	Ratio	Component   Registry Number
			T
0		x	17778-80-2
In	1	<b>x</b> .	7440-74-6
Na	1	x	7440-23-5
Mn	İ	X	7439-96-5
Li	1	×	7439-93-2

RETABLE

Referenced Author	Year   VOL	PG   R	eferenced Work	Referenced
(RAU)	(RPY) (RVL)	(RPG)	(RWK)	File
	=+=====+=====	=+=====+==		==+=======
Delnick Frank, M	1995	US	5426006 A	(HCAPLUS
Fujita Yuko	1998	WO	9804010 A	HCAPLUS
Nippon Telegr & Teleph	1989  013	JP	01-128354 A	HCAPLUS



```
Takeuchi Esther, S
                      |1996 |
                                   1
                                          |US 5558680 A
                                                               | HCAPLUS
L64 ANSWER 10 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN
AN
     1999:206870 HCAPLUS
DN
     130:252076
     Preparation of alcohol cyanoethyl ethers for lithium batteries
TI
     and organic electrolytic solutions containing them
ΙN
     Nishikawa, Satoshi
     Sunstar Engineering, Inc., Japan; Uni Sunstar Bv
PA
SO
     Jpn. Kokai Tokkyo Koho, 9 pp.
     CODEN: JKXXAF
DT
     Patent
     Japanese ·
LA
FAN.CNT 1
     PATENT NO.
                        KIND DATE
                                          APPLICATION NO.
                                                                  DATE
                              -----
                        ----
     JP 11080112
                        Α
                              19990326 JP 1997-245178
                                                                 19970910 <--
PRAI JP 1997-245178
                                19970910 <--
    MARPAT 130:252076
OS
     (R10) aR2 (OCH2CH2CN)b (I; R1 = C11-3 alkyl; R2 = residue of alcs. having
AB
     1-4 OH groups; a = 0-3; b = 1-4; a + b = 1-4) are prepared by
     cyanoethylation of alcs. with acrylonitrile (II) in the presence of
     ≥1 selected from (a) LiOH and (b) Li metal, Li alkoxides, compds.
     comprising Li and active methylene compds. such as Li acetylacetonate and
     in the absence of H2O. The organic electrolyte solns. for Li
     batteries or Li ion secondary batteries comprise I and
     Li salts dissolved therein. The electrolyte solns. may contain aprotic
     polar compds. This method gives I without discoloration due to polymerization
of
         II was added dropwise to a mixture of ethylene glycol and LiOH.H2O at
     40-0^{\circ} over 2 h, and the reaction mixture was further stirred at
     40-50° for 3 h to give ethylene glycol bis(2-cyanoethyl) ether
     (III) with purity ≥99.5%. LiClO4 was dissolved in III to give an
     electrolyte solution showing conductivity 2.7 + 10-3 S·Cm-1.
IC
     ICM C07C0255-13
     ICS B01J0023-04; C07C0253-30; H01M0010-40; C07B0061-00
CC
     23-19 (Aliphatic Compounds)
     Section cross-reference(s): 52
ST
     polyol cyanoethyl ether prepn battery electrolyte; alc
     cyanoethyl ether prepn battery electrolyte; acrylonitrile
     cyanoethylation polyol lithium hydroxide catalyst; ethylene glycol
     cyanoethyl ether battery electrolyte
IT
     Polar solvents.
     Polar solvents
        (aprotic; preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
ΙT
     Cyanoethylation
     Cyanoethylation
        (catalysts; preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
ΙT
     Ethylation catalysts
     Ethylation catalysts
        (cyanoethylation catalysts; preparation of (poly)alc. cyanoethyl ethers as
        battery electrolytes by LiOH-catalyzed reaction of polyols and
        acrylonitrile)
IT
     Primary batteries
     Secondary batteries
        (lithium; preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
    Alcohols, reactions
```

```
RL: RCT (Reactant); RACT (Reactant or reagent)
        (polyhydric; preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile),
ΙT
     Battery electrolytes
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
     Alcohols, reactions
     Glycols, reactions
     Polyoxyalkylenes, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
     7439-93-2, Lithium, uses 18115-70-3, Lithium
     acetylacetonate, uses
     RL: CAT (Catalyst use); USES (Uses)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT . 112-27-6P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
ΙT
     110-47-4P 110-67-8P, 2-Cyanoethyl methyl ether
     2141-62-0P
                  2465-91-0P
                               2465-93-2P
                                            3386-87-6P, Ethylene glycol
                                        9003-07-0DP, Polypropylene,
     bis(2-cyanoethyl) ether 6959-71-3P
     triol derivs., bis(2-cyanoethyl)ether 16792-83-9P, Propylene glycol
    bis(2-cyanoethyl) ether
                               22397-30-4P
                                             22397-31-5P, Diethylene glycol
     bis(2-cyanoethyl) ether
                               25265-71-8DP, Dipropylene glycol, ether with
     2-cyanoethyl and Me
                           35633-45-5P
                                         35633-50-2P
                                                       35633-51-3P
     39377-81-6P
                   39927-06-5P, Polyethylene glycol bis(2-cyanoethyl) ether
     51299-82-2P
                   57741-46-5P, Triethylene glycol bis(2-cyanoethyl)
     ether
             59113-36-9DP, Diglycerin, ether with tetrakis(2-cyanoethyl)
     61579-08-6P
                 180316-31-8P, 2,5,8,11-Tetraoxatetradecane-14-nitrile
     221628-60-0P
                    221628-62-2P
                                  221628-64-4P
     RL: IMF (Industrial manufacture); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
     56-81-5, 1,2,3-Propanetriol, reactions 57-55-6, 1,2-Propanediol,
     reactions
                64-17-5, Ethanol, reactions 67-56-1, Methanol, reactions
     67-63-0, Isopropanol, reactions 71-23-8, n-Propanol, reactions
     71-36-3, n-Butanol, reactions 102-71-6, Triethanolamine, reactions
     107-13-1, 2-Propenenitrile, reactions
                                            107-21-1, 1,2-Ethanediol,
     reactions
               109-86-4, Ethylene glycol monomethyl ether 110-80-5,
     Ethylene glycol monoethyl ether 111-46-6, Diethylene glycol, reactions
     111-77-3, Diethylene glycol monomethyl ether
                                                   112-35-6, Triethylene
     glycol monomethyl ether 115-77-5, reactions
                                                   122-20-3,
                          1320-67-8, Propylene glycol monomethyl ether
     Triisopropanolamine
     4439-20-7
                25265-71-8, Dipropylene glycol 25322-68-3
                                                             25322-69-4,
     Polypropylene glycol
                            25618-55-7 34590-94-8, Dipropylene glycol
     monomethyl ether 52125-53-8, Propylene glycol monoethyl ether
     59113-36-9, Diglycerin
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
    7791-03-9, Lithium perchlorate 14283-07-9
     RL: TEM (Technical or engineered material use); USES (Uses)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
ΙT
     1310-65-2, Lithium hydroxide
```

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RL: CAT (Catalyst use); USES (Uses)
        (preparation of polyol cyanoethyl ethers as battery electrolytes
        by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
     96-49-1, Ethylene carbonate
     RL: TEM (Technical or engineered material use); USES (Uses)
        (solvent; preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
IT
     7439-93-2, Lithium, uses 18115-70-3, Lithium
     acetylacetonate, uses
     RL: CAT (Catalyst use); USES (Uses)
        (preparation of (poly)alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
     7439-93-2 HCAPLUS
RN
CN
     Lithium (CA INDEX NAME)
Li
RN
     18115-70-3 HCAPLUS
CN
     2,4-Pentanedione, ion(1-), lithium (9CI) (CA INDEX NAME)
    Li<sup>+</sup>
     110-47-4P 110-67-8P, 2-Cyanoethyl methyl ether
     2141-62-0P 6959-71-3P 51299-82-2P
     RL: IMF (Industrial manufacture); SPN (Synthetic preparation); TEM
     (Technical or engineered material use); PREP (Preparation); USES (Uses)
        (preparation of (poly) alc. cyanoethyl ethers as battery
        electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)
RN
     110-47-4 HCAPLUS
CN
     Propanenitrile, 3-(1-methylethoxy)- (9CI) (CA INDEX NAME)
i-Pro-CH2-CH2-CN
RN
     110-67-8 HCAPLUS
CN
     Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)
                                                   fall group D
MeO-CH_2-CH_2-CN
     2141-62-0 HCAPLUS
RN
     Propanenitrile, 3-ethoxy- (9CI) (CA INDEX NAME)
CN
EtO-CH2-CH2-CN
RN 6959-71-3 HCAPLUS
```

CN Propanenitrile, 3-butoxy- (9CI) (CA INDEX NAME)

n-BuO-CH2-CH2-CN

RN 51299-82-2 HCAPLUS CN Propanenitrile, 3-propoxy- (9CI) ( group D

1,372,1033

n-Pro-CH2-CH2-CN

● Li

RN 14283-07-9 HCAPLUS CN Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

• Li+

 ${\tt Li-OH}$ 

IT 96-49-1, Ethylene carbonate

RL: TEM (Technical or engineered material use); USES (Uses) (solvent; preparation of (poly)alc. cyanoethyl ethers as **battery** electrolytes by LiOH-catalyzed reaction of polyols and acrylonitrile)

RN 96-49-1 HCAPLUS

CN . 1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)

L64 ANSWER 11 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1997:129574 HCAPLUS

DN 126:133588

TI Nonaqueous electrolyte **batteries** using electrolytes containing self discharge inhibitors

IN Jinno, Maruo; Uehara, Mayumi; Sakurai, Atsushi; Nishio, Koji; Saito, Toshihiko

PA Sanyo Denki Kk, Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.		DATE	APPLICATION NO.	DATE.	
PI	JP 08321312	Α	19961203	JP 1995-150844	19950524 <	
PRAI	JP 1995-150844		19950524	< ·		

Li batteries use electrolytes containing LiCF3SO3 or LiPF6 dissolved in high dielec. constant solvent selected from ethylene carbonate, propylene carbonate, and butylene carbonate; where the electrolytes contain 1-20 volume% additive selected from triethylamine, n-butylamine, aniline, tri-Me hydroxylamine, 1-dimethylamino-2-methoxy ethane, acetonitrile, acrylonitrile, 3-methoxy propionitrile, benzonitrile, nitromethane, nitroethane, N,N-dimethylacetamide, N,N-dimethylformamide, formamide, N-methyl-2-pyrrolidone, N, N'-dimethyl imidazolidinone, isoxazole, 3,5-di-Me isoxazole, 3-methyl-2-oxazolidone, 1,2,3-oxadiazole, N-Me morpholine, di-Me sulfide, Et Me sulfide, 2-Me thiophene, 1-butane thiol, benezenethiol, di-Me sulfate, di-Et sulfate, di-Me sulfite, di-Et sulfite, butadienesulfone, 3-Me sulfolene, 1,4-thioxane, phenoxathiin, 1,4-thiazine, thiomorpholine, pyridine, 1,3-dimethyl-2-imidazolidinone, DMSO, di-Me sulfone, Me Et sulfonate, and di-Me sulfinite. The electrolytes may contain 1,2-dimethoxyethane. Since the additives react with Li in anodes and the solvents and the solutes in the electrolytes to form coatings on the anodes for prevention of the reaction between the electrolytes and the anodes, the batteries have improved storage property. These batteries have long shelf life.

IC ICM H01M0006-16

ICS H01M0010-40

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST lithium battery electrolyte self discharge inhibitor

IT Battery electrolytes

(self discharge inhibitors in nonaq. electrolyte solns. for lithium batteries)

TT 7439-93-2, Lithium, uses 21324-40-3, Lithium
hexafluorophosphate 33454-82-9, Lithium
trifluoromethanesulfonate
RL: DEV (Device component use); USES (Uses)

```
(nonag. electrolyte solns. containing self discharge inhibitors for lithium
        batteries)
     62-53-3, Aniline, uses
ΙT
                               64-67-5, Diethyl sulfate
                                                            67-68-5,
     Dimethylsulfoxide, uses 67-71-0, Dimethylsulfone 68-12-2,
     N, N-Dimethylformamide, uses 75-05-8, Acetonitrile, uses
                                                                    75-12-7,
     Formamide, uses 75-18-3, Dimethylsulfide
                                                   75-52-5, Nitromethane, uses
     77-78-1, Dimethyl sulfate
                                 79-24-3, Nitroethane
                                                           80 - 73 - 9,
     N,N'-Dimethylimidazolidinone 100-47-0, Benzonitrile, uses
                                                                      107-13-1,
                           108-98-5, Benzenethiol, uses
     Acrylonitrile, uses
                                                            109-02-4,
     N-Methylmorpholine 109-73-9, n-Butylamine, uses
                                                            109-79-5,
     1-Butanethiol 110-67-8, 3-Methoxypropionitrile 110-86-1,
     Pyridine, uses 121-44-8, Triethylamine, uses
                                                       123-90-0, Thiomorpholine
     127-19-5, N,N-Dimethylacetamide 262-20-4, Phenoxathiin
                                                                   288-14-2,
     Isoxazole 288-43-7, 1,2,3-Oxadiazole 290-56-2, 1,4-Thiazine
     290-57-3, 1,4-Thiazine 300-87-8, 3,5-Dimethylisoxazole 554-14-3, 2-Methylthiophene 616-42-2, Dimethyl sulfite 623-81-4, Diethyl sulfite
     624-89-5, Ethylmethylsulfide 666-15-9 872-50-4, N-Methyl-2-pyrrolidone, uses 1193-10-8, 3-Methylsulfolene 1912-28-3, Methyl ethyl
     sulfonate 3030-44-2
                              5669-39-6, Trimethylhydroxylamine
                                                                    15980-15-1,
     1,4-Thioxane
                    19836-78-3 28452-93-9, Butadienesulfone
     RL: DEV (Device component use); MOA (Modifier or additive use); USES
     (Uses)
         (self discharge inhibitors in nonaq, electrolyte solns, for lithium
        batteries)
IT
     96-49-1, Ethylene carbonate 108-32-7, Propylene
                 110-71-4, 1,2-Dimethoxyethane 4437-85-8, Butylene
     carbonate
     RL: DEV (Device component use); USES (Uses)
         (solvents for nonag. electrolyte solns. containing self discharge
        inhibitors for lithium batteries)
IT
     7439-93-2, Lithium, uses 21324-40-3, Lithium
     hexafluorophosphate 33454-82-9, Lithium
     trifluoromethanesulfonate
     RL: DEV (Device component use); USES (Uses)
        (nonaq. electrolyte solns. containing self discharge inhibitors for lithium
        batteries)
RN
     7439-93-2 HCAPLUS
     Lithium (CA INDEX NAME)
CN
Li
     21324-40-3 HCAPLUS
RN
CN
     Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)
```

33454-82-9 HCAPLUS RN CN Methanesulfonic acid, trifluoro-, lithium salt (8CI, 9CI) (CA INDEX NAME)

IT 110-67-8, 3-Methoxypropionitrile RL: DEV (Device component use); MOA (Modifier or additive use); USES (self discharge inhibitors in nonaq. electrolyte solns. for lithium batteries) RN 110-67-8 HCAPLUS

CN Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)

MeO-CH2-CH2-CN

IT 96-49-1, Ethylene carbonate 108-32-7, Propylene carbonate 4437-85-8, Butylene carbonate RL: DEV (Device component use); USES (Uses) (solvents for nonaq. electrolyte solns. containing self discharge inhibitors for lithium batteries)

96-49-1 HCAPLUS RN

1,3-Dioxolan-2-one (9CI) (CA INDEX NAME)

108-32-7 HCAPLUS 1,3-Dioxolan-2-one, 4-methyl- (9CI) (CA INDEX NAME)

jan delaval - 27 february 2007

RN 4437-85-8 HCAPLUS 1,3-Dioxolan-2-one, 4-ethyl- (9CI) (CA INDEX NAME)

L64 ANSWER 12 OF 12 HCAPLUS COPYRIGHT 2007 ACS on STN

ΑN 1989:176598 HCAPLUS

DN 110:176598

ΤI Research of new solvents for lithium batteries. II. Behavior of aliphatic nitriles substituted by electron donating groups

ΑU Guibert, Sylvie; Cariou, Michel; Simonet, Jacques

CS Lab. Electrochim., Univ. Rennes I, Rennes, 35042, Fr.

SO Bulletin de la Societe Chimique de France (1988), (6), 924-9 CODEN: BSCFAS; ISSN: 0037-8968

DT Journal

LA French

AB Methoxyacetonitrile (I), methoxy-3-propionitrile, and cyano-1-pyrrolidine have low reactivity towards Li and a broad electrochem. stability window (>5.0 V), suitable for use as electrolyte solvents in Li batteries The dimer of I is formed in basic conditions or in the presence of an alkali metal and can be oxidized electrochem. At potentials more cathodic than the equilibrium potential of the Li/Li+ couple, I is electrochem. unstable, but the formation of the dimer can be reversed by controlling the pH of the medium.

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 28, 72

ST methoxyacetonitrile stability electrolyte lithium battery; cyanopyrrolidine stability electrolyte lithium battery; methoxypropionitrile stability electrolyte lithium battery

IT Batteries, secondary

(lithium, electrolytes for, aliphatic nitrile solvents for, stability of)

ΙT 110-67-8, Methoxy-3-propionitrile 1530-88-7, Cyano-1-pyrrolidine 1738-36-9, Methoxyacetonitrile

RL: USES (Uses)

(chemical and electrochem. stability of, for lithium battery electrolyte use)

29935-35-1, Lithium hexafluoroarsenate (LiAsF6) IT RL: USES (Uses)

(electrolytes containing, aliphatic nitrile solvents for, stability of)

IT 7439-93-2, Lithium, reactions

> RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with aliphatic nitrile solvents, battery electrolyte use in relation to)

IT 110-67-8, Methoxy-3-propionitrile 1738-36-9,

Methoxyacetonitrile

RL: USES (Uses)

(chemical and electrochem. stability of, for lithium battery electrolyte use)

110-67-8 HCAPLUS RN

CN Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)

 $MeO-CH_2-CH_2-CN$ 

group I

RN 1738-36-9 HCAPLUS

Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME) CN

 $H_3C-O-CH_2-C \equiv N$ 

IT 29935-35-1, Lithium hexafluoroarsena'ce ( Lithium ) RL: USES (Uses)

(electrolytes containing, aliphatic nitrile solvents for, stability of)

RN 29935-35-1 HCAPLUS

Arsenate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME) CN

● Li +

ΙT 7439-93-2, Lithium, reactions

RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with aliphatic nitrile solvents, battery electrolyte use in relation to)

RN 7439-93-2 HCAPLUS

CN Lithium (CA INDEX NAME)

Li

=> d his

(FILE 'HOME' ENTERED AT 06:22:08 ON 27 FEB 2007) SET COST OFF

FILE 'REGISTRY' ENTERED AT 06:22:38 ON 27 FEB 2007 ACT LAURA731B/A

27 SEA FILE=REGISTRY ABB=ON PLU=ON (463-79-6/BI OR 1001-55-4/BI L1

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ACT LAURA731/A
L2
                 STR
                 SCR 2043 OR 1838 OR 1993 OR 2021 OR 2016 OR 2026 OR 1918 2039 O
L3
L4
             436 SEA FILE=REGISTRY CSS FUL L2 NOT L3
                ACT LAURA731A/A
                _____
L5
                SCR 2043 OR 1838 OR 1993 OR 2021 OR 2016 OR 2026 OR 1918 2039 O
L6
L7
              55 SEA FILE=REGISTRY CSS FUL L6 NOT L5
               _____
rs
             10 S L1 AND (NITRILE OR CYANO)
L9
              9 S L8 AND 1/N
L10
             498 S L4, L7, L9
L11
                STR L2
L12
                STR L6
             40 S L11 OR L12 NOT L3 CSS SAM
L13
                SCR 2043 OR 1838 OR 1993 OR 2021 OR 2016 OR 2026 OR 1918 OR 203
L14
L15
             40 S (L11 OR L12) NOT L14 CSS SAM
L16
             724 S (L11 OR L12) NOT L14 CSS FUL
                 SAV LAURA731C/A L16 TEMP
L17
             732 S L9, L10, L16
. L18
             727 S L17 NOT ((D OR T)/ELS OR 11C# OR 13C# OR 14C# OR C11# OR C13#
     FILE 'HCAPLUS' ENTERED AT 06:49:30 ON 27 FEB 2007
L19
           1653 S L18
     FILE 'REGISTRY' ENTERED AT 06:49:47 ON 27 FEB 2007
               5 S 7791-03-9 OR 14283-07-9 OR 21324-40-3 OR 29935-35-1 OR 90076-
L20
L21
               6 S LI/MF NOT MASS
               5 S 12031-65-1 OR 12057-17-9 OR 12190-79-3 OR 15365-14-7 OR 31134
L22
          77211 S (LI/ELS OR ?LITHIUM?/CNS OR 7439-93-2/CRN) AND (O/ELS OR 1777
L23
L24
           6981 S L23 AND (NI/ELS OR ?NICKEL?/CNS OR 7440-02-0/CRN)
            6839 S L23 AND (CO/ELS OR ?COBALT?/CNS OR 7440-48-4/CRN)
L25
           8874 S L23 AND (MN/ELS OR ?MANGANSESE?/CNS OR 7439-96-5/CRN)
L26
           1501 S L24-L26 AND 3/ELC.SUB
L27
            428 S L24 AND L25 AND 4/ELC.SUB
L28
L29
           1935 S L27, L28, L20, L22
     FILE 'HCAPLUS' ENTERED AT 06:54:48 ON 27 FEB 2007
L30
            32 S L29 AND L19
L31
             14 S L21 AND L19
L32
             38 S L30, L31
     FILE 'REGISTRY' ENTERED AT 06:55:25 ON 27 FEB 2007
              0 S L1 AND LI/ELS NOT L29, L21
L33
               5 S 96-49-1 OR 105-58-8 OR 108-32-7 OR 616-38-6 OR 623-53-0
L34
     FILE 'HCAPLUS' ENTERED AT 06:56:20 ON 27 FEB 2007
              17 S L34 AND L32
L35
              6 S L32 AND PY<=2003 NOT P/DT
L36
              17 S L32 AND (PD<=20031209 OR PRD<=20031209 OR AD<=20031209) AND P
L37
L38
              23 S L36, L37
                E SUN/AU
                E SUN L/AU
             297 S E3
L39
L40
             17 S E28
                E SUN LU/AU
             145 S E3
L41
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3 S E15
L42
             17 S E50
L43
               E SUN NAME/AU
L44
             11 S E4
               E LU/AU
L45
              5 S E3
             E LU Y/AU
L46
            734 S E3
L47
            272 S E24
               E LU YING/AU
L48
            357 S E3
               E LU NAME/AU
L49
              2 S E4
              E LUYING/AU
L50
             1 S L39-L49 AND L19
L51
             23 S L38, L50
            11 S L51 NOT BATTERY
L52
L53
            12 S L51 NOT L52
                SEL RN
    FILE 'REGISTRY' ENTERED AT 07:01:38 ON 27 FEB 2007
L54
           189 S E1-E189
L55
            14 S L54 AND L18
L56
             14 S L54 AND L29
L57
             12 S L54 AND LI/ELS NOT L56
L58
             2 S L54 AND C/MF
L59
             5 S L54 AND L34
L60
            2 S L54 AND OCOC2/ES NOT L59
L61
            140 S L54 NOT L55-L60
L62
             1 S L61 AND CH2O3
     FILE 'HCAPLUS' ENTERED AT 07:11:16 ON 27 FEB 2007
L63
            12 S L53 AND L55-L60,L62
L64
             12 S L63 AND BATTER?/CW,CT,BI
     FILE 'REGISTRY' ENTERED AT 07:12:30 ON 27 FEB 2007
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FILE 'HCAPLUS' ENTERED AT 07:12:46 ON 27 FEB 2007

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jan delaval - 27 february 2007

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=> d his
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(FILE 'HOME' ENTERED AT 07:38:39 ON 27 FEB 2007)
                 SET COST OFF
     FILE 'REGISTRY' ENTERED AT 07:38:49 ON 27 FEB 2007
L1
               1 S 14283-07-9
L2
               1 S 21324-40-3
L3
           63030 S 16919-18-9/CRN
L4
           62733 S 14874-70-5/CRN
L5
              72 S L3 AND L4
L6
              0 S L5 AND LI/ELS
L7
              56 S L3 AND LI/ELS
             110 S L4 AND LI/ELS
rs
L9
              38 S L7 AND 2/NC
L10
              56 S L8 AND 2/NC
              31 S L9 AND NR>=1
L11
L12
              7 S L9 NOT L11
L13
              81 S L8 AND NR>=1
              29 S L8 NOT L13
L14
     FILE 'HCAPLUS' ENTERED AT 07:42:47 ON 27 FEB 2007
           3382 S L1
L15
            2279 S (LI OR LITHIUM) () TETRAFLUOROBORATE
L16
L17
               5 S (LI OR LITHIUM) () TETRAFLUORO BORATE
L18
               O S (LI OR LITHIUM) () TETRA FLUORO BORATE
              1 S (LI OR LITHIUM) () TETRA FLUOROBORATE
L19
L20
             664 S (LI OR LITHIUM) () FLUOROBORATE
               2 S (LI OR LITHIUM) () FLUORO BORATE
L21
            1876 S LIBF4
L22
L23
             10 S LI BF4
            4157 S L15-L23
L24
L25
            5309 S L2
            5022 S (LI OR LITHIUM) () HEXAFLUOROPHOSPHATE
L26
L27
              21 S (LI OR LITHIUM) () HEXAFLUORO PHOSPHATE
L28
               O S (LI OR LITHIUM) () HEXA FLUORO PHOSPHATE
L29
               3 S (LI OR LITHIUM) () HEXA FLUOROPHOSPHATE
L30
             103 S (LI OR LITHIUM) () FLUOROPHOSPHATE
L31
               O S (LI OR LITHIUM) () FLUORO PHOSPHATE
.L32
            2387 S LIPF6 OR LI PF6
            5966 S L25-L32
L33
L34
           1647 S L24 AND L33
L35
             246 S L34 AND PY<=2003 NOT P/DT
           1158 S L34 AND (PD<=20031209 OR PRD<=20031209 OR AD<=20031209) AND P
L36
           1404 S L35, L36
L37
           1244 S L37 AND BATTERY
L38
                 E BATTERIES/CT
                 E E3+ALL
             840 S L37 AND (E1 OR E2+OLD, NT OR E3+OLD, NT OR E4+OLD, NT OR E5+OLD,
L39
                 E BATTERY/CT
             997 S L37 AND (E4+OLD,NT OR E5+OLD,NT OR E6+OLD,NT OR E7 OR E8+OLD.
L40
                 E E9+ALL
              48 S L37 AND (E2+OLD, NT OR E3+OLD, NT OR E4+OLD, NT)
L41
L42
           1083 S L37 AND H01M/IPC, IC, ICM, ICS
L43
            1270 S L38-L42
L44
             438 S L43 AND (MIX? OR BLEND?)
     FILE 'REGISTRY' ENTERED AT 07:48:11 ON 27 FEB 2007
                ACT LAURA731C/A
```

\_\_\_\_\_

www.bna.co

L45 STR
L46 STR
L47 SCR 2043 OR 1838 OR 1993 OR 2021 OR 2016 OR 2026 OR 1918 OR 203
L48 724 SEA FILE=REGISTRY CSS FUL (L45 OR L46) NOT L47

FILE 'HCAPLUS' ENTERED AT 07:48:30 ON 27 FEB 2007

L49 3 S L48 AND L43 L50 1 S L44 AND L49 L51 3 S L49,L50

=> fil hcaplus

FILE 'HCAPLUS' ENTERED AT 07:49:05 ON 27 FEB 2007 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS)

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FILE COVERS 1907 - 27 Feb 2007 VOL 146 ISS 10 FILE LAST UPDATED: 26 Feb 2007 (20070226/ED)

Claim Z

New CAS Information Use Policies, enter HELP USA

This file contains CAS Registry Numbers for eas substance identification.

=> d 151 all hitstr retable tot

L51 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 2005:497321 HCAPLUS

DN 143:29529

ED Entered STN: 10 Jun 2005

TI Nonaqueous electrolytes having an extended temperature range for **battery** applications

IN Sun, Luying

PA USA

SO U.S. Pat. Appl. Publ., 17 pp. CODEN: USXXCO

DT Patent

LA English

IC ICM H01M0010-40

ICS H01M0004-52; H01M0004-50; H01M0004-58

INCL 429326000; 429330000; 429339000; 429231300; 429231100; 429223000; 429221000; 429224000; 429231800

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
Section cross-reference(s): 72, 76

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE	
ΡI	US 2005123835	A1	20050609	US 2003-731268	20031209 <	

```
PRAI US 2003-731268
                                 20031209 <--
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
. PATENT NO.
 US 2005123835
                 ICM
                         H01M0010-40
                         H01M0004-52; H01M0004-50;
                 ICS
                         H01M0004-58
                 INCL
                         429326000; 429330000; 429339000; 429231300; 429231100;
                         429223000; 429221000; 429224000; 429231800
                 IPCI
                         H01M0010-40 [ICM, 7]; H01M0010-36 [ICM, 7, C*];
                         H01M0004-52 [ICS,7]; H01M0004-50 [ICS,7]; H01M0004-58
                         [ICS, 7]
                                                                               <--
                         H01M0010-36 [I,C*]; H01M0010-40 [I,A]
                 IPCR
                                                                               <--
                         429/326.000; 429/221.000; 429/223.000; 429/224.000;
                 NCL
                         429/231.100; 429/231.300; 429/231.800; 429/330.000;
                         429/339.000
                         H01M010/40E1
                  ECLA
     MARPAT 143:29529
OS
AB
     The present invention discloses non-aqueous electrolytes having an extended
     temperature range for battery applications. The electrolyte comprises
     an electrolyte salt, e.g., LiPF6, a first non-aqueous solvent, and a
     second non-aqueous solvent. The electrolyte of the present invention has
     higher ionic conductivity, lower f.p., and lower vapor pressure at high
temperature
     than com. electrolytes. These non-aqueous electrolytes can be used, for
     example, in lithium-ion batteries. Methods of making
     lithium-ion batteries are also described.
ST
     battery nonaq electrolyte extended temp range
IT
     Electrochromic devices
     Sensors
        (electrolyte; nonaq. electrolytes having extended temperature range for
        battery applications)
IT
     Secondary batteries
        (lithlum; nonaq. electrolytes having extended temperature range for
        battery applications)
IT
     Battery electrolytes
     Electrolytic capacitors
       Fuel cell electrolytes
     Ionic conductivity
        (nonaq. electrolytes having extended temperature range for battery
        applications)
     Carbonaceous materials (technological products)
IT
     Coke
     Esters, uses
     Éthers, uses
     RL: DEV (Device component use); USES (Uses)
        (nonaq. electrolytes having extended temperature range for battery
        applications)
IT
     Sulfonic acids, uses
     RL: DEV (Device component use); USES (Uses)
        (perfluoro, lithium salt; nonaq. electrolytes having extended temperature
        range for battery applications)
ΙT
     Perfluoro compounds
     RL: DEV (Device component use); USES (Uses)
        (sulfonic acids, lithium salt; nonaq. electrolytes having extended
        temperature range for battery applications)
     96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 110-67-8, 3-Methoxypropionitrile
ΙT
     463-79-6D, Carbonic acid, ester, cyclic 463-79-6D, Carbonic acid, ester,
     linear 616-38-6, Dimethyl carbonate 623-53-0, Ethyl methyl carbonate
```

```
1001-55-4, 2-Acetoxyacetonitrile 1656-48-0 1738-36-9,
     Methoxyacetonitrile 2141-62-0, 3-Ethoxypropionitrile
     7782-42-5, Graphite, uses 7791-03-9, Lithium perchlorate 12031-65-1,
     Lithium nickel oxide (LiNiO2) 12057-17-9, Lithium manganese oxide
                 12190-79-3, Cobalt lithium oxide (CoLiO2) 14283-07-9
     (LiMn2O4)
     , Lithium tetrafluoroborate 15365-14-7, Iron lithium
     phosphate felipo4 18804-04-1, uses 21324-40-3,
     Lithium hexafluorophosphate 29935-35-1, Lithium
     hexafluoroarsenate 56756-91-3 62957-60-2,
     Ethoxyacetonitrile 90076-65-6 260362-83-2
                                                     311346-25-5, Cobalt
     lithium nickel oxide (Co0.1-0.9LiNi0.1-0.902) 852995-04-1
     RL: DEV (Device component use); USES (Uses)
        (nonag. electrolytes having extended temperature range for battery
        applications)
ΙT
     110-67-8, 3-Methoxypropionitrile 1001-55-4,
     2-Acetoxyacetonitrile 1738-36-9, Methoxyacetonitrile
     2141-62-0, 3-Ethoxypropionitrile 14283-07-9,
     Lithium tetrafluoroborate 18804-04-1, uses
     21324-40-3, Lithium hexafluorophosphate
     56756-91-3 62957-60-2, Ethoxyacetonitrile
     852995-04-1
     RL: DEV (Device component use); USES (Uses)
        (nonag. electrolytes having extended temperature range for battery
        applications)
RN
     110-67-8 HCAPLUS
     Propanenitrile, 3-methoxy- (9CI) (CA INDEX NAME)
CN
MeO-CH_2-CH_2-CN
     1001-55-4 HCAPLUS
RN
     Acetonitrile, (acetyloxy) - (9CI) (CA INDEX NAME)
CN
Aco-CH2-CN
     1738-36-9 HCAPLUS
RN
     Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)
CN
H_3C-O-CH_2-C \equiv N
     2141-62-0 HCAPLUS
RN
     Propanenitrile, 3-ethoxy- (9CI) (CA INDEX NAME)
CN
EtO-CH_2-CH_2-CN
     14283-07-9 HCAPLUS
RN
     Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)
CN
```

● T.i f

RN 18804-04-1 HCAPLUS CN Carbonic acid, 1-cyano-1-methylethyl methyl ester (9CI) (CA INDEX NAME)

RN 21324-40-3 HCAPLUS CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

T.4 ↑

RN 56756-91-3 HCAPLUS CN Propanenitrile, 2-(acetyloxy)-2-methyl- (9CI) (CA INDEX NAME)

RN 62957-60-2 HCAPLUS CN Acetonitrile, ethoxy- (6CI, 9CI) (CA INDEX NAME)

jan delaval - 27 february 2007

Eto-CH2-CN

```
RN
    852995-04-1 HCAPLUS
    Carbonic acid, cyanomethyl methyl ester (9CI) (CA INDEX NAME)
CN
MeO-C-O-CH2-CN
L51 ANSWER 2 OF 3 HCAPLUS COPYRIGHT 2007 ACS on STN
    2000:399134 HCAPLUS
    133:20143
    Entered STN: 16 Jun 2000
    Nonaqueous electrolyte solutions and secondary nonaqueous electrolyte
    batteries
    Hayashi, Takeshi; Tan, Hiroaki
ΙN
    Mitsui Petrochemical Industries, Co., Ltd., Japan
    Jpn. Kokai Tokkyo Koho, 7 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IÇ
    ICM H01M0010-40
    52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
CC
FAN.CNT 1
    PATENT NO.
                        KIND
                               DATE
                                         APPLICATION NO.
                                                                 DATE
                        ____
                                          ______
                                                                 -----
     JP 2000164249
                        A
                                          JP 1998-336632
                               20000616
                                                                 19981127 <--
PRAI JP 1998-336632
                               19981127 <--
CLASS
               CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _____
 JP 2000164249
                ICM
                       H01M0010-40
                       H01M0010-40 [ICM, 7]
                IPCI
                                                                          <--
                       H01M0010-36 [I,C*]; H01M0010-40 [I,A]
                IPCR
                                                                          <--
OS
    MARPAT 133:20143
    The electrolyte solns. contain an electrolyte and a nonaq. solvent containing
AΒ
     a F-containing cyanoethyl ether X(OR)n)CH2CH2CN, where X = F substituted C1-10
    hydrocarbon group, R = C2-4 alkylene group, and n = 0-30. The salt is
     selected from LiPF6, LiBF4, and Li salts of S containing
     organic acids. The batteries are secondary Li batteries.
ST
    secondary lithium battery electrolyte solvent; lithium
    battery electrolyte solvent fluorinated cyanoethyl ether
IT
    Battery electrolytes
        (nonag. solvent mixts. containing fluorinated cyanoethyl ethers
       for electrolytes in secondary lithium batteries)
IT
     96-49-1, Ethylene carbonate 353-18-4 616-38-6, Dimethyl
     carbonate 21324-40-3, Lithium
     hexafluorophosphate 85169-02-4 272128-06-0
     272128-07-1 272128-08-2 272128-09-3 272128-10-6
     RL: DEV (Device component use); USES (Uses)
        (nonaq. solvent mixts. containing fluorinated cyanoethyl ethers
       for electrolytes in secondary lithium batteries)
IT
     353-18-4 21324-40-3, Lithium
    hexafluorophosphate 272128-06-0
     RL: DEV (Device component use); USES (Uses)
```

(nonaq. solvent mixts. containing fluorinated cyanoethyl ethers for electrolytes in secondary lithium batteries)

RN 353-18-4 HCAPLUS

CN Propanenitrile, 3-(2-fluoroethoxy)- (9CI) (CA INDEX NAME)

FCH2-CH2-O-CH2-CH2-CN

RN 21324-40-3 HCAPLUS

CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)

• Li +

RN 272128-06-0 HCAPLUS

CN Propanenitrile, 3-(2,2,2-trifluoroethoxy)- (9CI) (CA INDEX NAME)

F3C-CH2-O-CH2-CH2-CN

L51 ANSWER 3 OF 3 HCAPLUS COPYRIGHT 2007 ACS on STN

AN 1999:737130 HCAPLUS

DN 131:325078

ED Entered STN: 19 Nov 1999

TI Primary or secondary electrochemical generator

IN Gratzel, Michael; Sugnaux, Francois R.; Pappas, Nicholas

PA Ecole Polytechnique Federale De Lausanne (Epfl) Sri, Switz.

SO PCT Int. Appl., 29 pp. CODEN: PIXXD2

DT Patent

LA English

IC ICM **H01M0010-40** 

ICS H01M0004-48; H01M0004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

FAN.CNT 1

PATENT NO. KIND DATE APPLICATION NO. DATE

PI WO 9959218 A1 19991118 WO 1999-EP3261 19990508 <-W: CN, JP, US

RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,

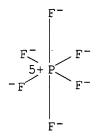
PT, SE
EP 1086506 A1 20010328 EP 1999-932452 19990508 <-R: CH, DE, FR, GB, LI, NL, IE

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PRAI EP 1998-810431
                                 19980512
                          Α
     WO 1999-EP3261
                                 19990508
                                          <--
CLASS
              CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
WO 9959218
                 ICM
                        H01M0010-40
                        H01M0004-48; H01M0004-58
                 ICS
                 IPCI
                        H01M0010-40 [ICM, 6]; H01M0010-36 [ICM, 6, C*];
                        H01M0004-48 [ICS, 6]; H01M0004-58 [ICS, 6]
                                                                               <--
                 IPCR
                        H01M0004-48 [I,C*]; H01M0004-48 [I,A]; H01M0004-58
                        [I,C*]; H01M0004-58 [I,A]; H01M0010-36 [I,C*];
                        H01M0010-40 [I,A]
                                                                               <---
                 ECLA
                        H01M004/48; H01M004/48B2; H01M004/58B; H01M010/40L
 EP 1086506
                 IPCI
                        H01M0010-40 [ICM, 6]; H01M0010-36 [ICM, 6, C*];
                        H01M0004-48 [ICS, 6]; H01M0004-58 [ICS, 6]
                 IPCR
                        H01M0004-48 [I,C*]; H01M0004-48 [I,A]; H01M0004-58
                        [I,C^*]; HO1MO004-58 [I,A]; HO1MO010-36 [I,C^*];
                        H01M0010-40 [I,A]
     A high power d. and high capacity primary or secondary electrochem.
     generator has at least one electrode composed of an elec. active solid
     material, the electrode having a mesoporous texture forming a
     bi-continuous junction of large sp. surface area with the electrolyte.
     The specific morphol. of the electroactive material permits high rates of
     ion insertion in the solid while allowing for rapid ion transport in
     electrolyte present in the porous space of the electrode. Specific
     methods for preparation of the electrode are disclosed, in particular the
     control of the electrode morphol. by use of surfactant assemblies such as
     surfactant micelles exerting a templating effect during the chemical
     synthesis of the electroactive material.
ST
     battery electrode transition metal oxide chalcogenide
IT.
     Primary batteries
       Secondary batteries
        (lithium; primary or secondary electrochem. generator)
IT
     Battery electrodes
        (primary or secondary electrochem. generator)
IT
     Transition metal chalcogenides
     Transition metal oxides
     RL: DEV (Device component use); USES (Uses)
        (primary or secondary electrochem. generator)
IT
     Glass, uses
     RL: DEV (Device component use); USES (Uses)
        (separator; primary or secondary electrochem. generator)
ΙT
     Titanium alloy
     RL: DEV (Device component use); USES (Uses)
        (current collector; primary or secondary electrochem. generator)
IT
     Aluminum alloy
     RL: DEV (Device component use); USES (Uses)
        (primary or secondary electrochem. generator)
IT
     7440-44-0, Carbon, uses 12597-68-1, Stainless steel, uses
     RL: DEV (Device component use); USES (Uses)
        (current collector; primary or secondary electrochem. generator)
IT
     7782-42-5, Graphite, uses
     RL: DEV (Device component use); USES (Uses)
        (paper, current collector; primary or secondary electrochem. generator)
     96-48-0 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate
     108-32-7, Propylene carbonate 616-38-6, Dimethyl carbonate
                 1309-37-1, Iron oxide (Fe2O3), uses 1312-43-2, Indium oxide Manganese dioxide, uses 1313-27-5, Molybdenum trioxide, uses
     Dioxolane
     1313-13-9, Manganese dioxide, uses
     1313-96-8, Niobium pentoxide 1314-35-8, Tungsten trioxide, uses
     1314-62-1, Vanadium pentoxide, uses
                                           1317-33-5, Molybdenum sulfide mos2,
```

```
1317-61-9, Iron oxide (Fe304), uses 1738-36-9,
     Methoxyacetonitrile 2923-17-3, Lithium trifluoroacetate 11113-84-1,
     Ruthenium oxide 11126-12-8, Iron sulfide 11129-18-3, Cerium oxide 12039-13-3, Titanium disulfide 12055-23-1, Hafnium dioxide 12067-45-7, Titanium diselenide 12138-09-9, Tungsten sulfide ws2 12645-46-4,
                     13463-67-7, Titania, uses
     Iridium oxide
                                                    14024-11-4, Lithium
     tetrachloroaluminate 14283-07-9, Lithium
     tetrafluoroborate
                          18424-17-4, Lithium hexafluoroantimonate
     21324-40-3, Lithium hexafluorophosphate 28856-69-9, Methoxypropionitrile 28106-65-2, Tetrafluoropropanol
     29935-35-1, Lithium hexafluoroarsenate 33454-82-9, Lithium triflate
     37245-92-4, Ruthenium sulfide 39300-70-4, Lithium nickel oxide
                                            52627-24-4, Cobalt lithium oxide
     39457-42-6, Lithium manganese oxide
     59763-75-6, Tantalum oxide 66216-18-0 90076-65-6 131344-56-4, Cobalt
     lithium nickel oxide 131651-65-5, 1-Butanesulfonic acid,
     1,1,2,2,3,3,4,4,4-nonafluoro-, lithium salt 132404-42-3 248588-09-2,
     Indium lithium manganese sodium oxide
     RL: DEV (Device component use); USES (Uses)
         (primary or secondary electrochem. generator)
     82113-65-3D, alkylguanidinium compound 145022-44-2 174899-82-2
TΤ
     RL: DEV (Device component use); MOA (Modifier or additive use); USES
     (Uses)
         (primary or secondary electrochem. generator)
     1314-23-4, Zirconia, uses 1344-28-1, Aluminum oxide (Al2O3), uses
TΤ
     9003-07-0, Polypropylene
     RL: DEV (Device component use); USES (Uses)
        (separator; primary or secondary electrochem. generator)
RE.CNT
              THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Delnick Frank, M; US 5426006 A 1995 HCAPLUS
(2) Fujita Yuko; WO 9804010 A 1998 HCAPLUS
(3) Nippon Telegr & Teleph Corp; JP 01-128354 A 1989, V013(374) HCAPLUS
(4) Takeuchi Esther, S; US 5558680 A 1996 HCAPLUS
     1738-36-9, Methoxyacetonitrile 14283-07-9,
     Lithium tetrafluoroborate 21324-40-3,
     Lithium hexafluorophosphate
     RL: DEV (Device component use); USES (Uses)
         (primary or secondary electrochem. generator)
RN
     1738-36-9 HCAPLUS
     Acetonitrile, methoxy- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)
CN
                                       group A
H_3C-O-CH_2-C \equiv N
RN
     14283-07-9 HCAPLUS
     Borate(1-), tetrafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)
```

● Li+

RN 21324-40-3 HCAPLUS CN Phosphate(1-), hexafluoro-, lithium (8CI, 9CI) (CA INDEX NAME)



## ● Li+

## RETABLE

Referenced Author (RAU)	Year   VOL  (RPY) (RVL)	(RPG)	eferenced Work (RWK)	Referenced   File
Delnick Frank, M	1995		5426006 A	HCAPLUS
Fujita Yuko	1998	WO	9804010 A	HCAPLUS
Nippon Telegr & Teleph	1989  013	JP	01-128354 A	HCAPLUS
Takeuchi Esther, S	1996	US	5558680 A	HCAPLUS